

DFT Study in Strain Engineering HfSe₂ Nanosheets for Sensing SO₂, SOF₂, and SO₂F₂ Gases

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Abstract—Strain engineering is a technique of tuning a material's properties by modifying its mechanical or structural characteristics. Using first-principles calculations, the strain impact on the electronic structure of layered HfSe₂ nanosheets has been studied to examine the sensing capabilities of the layered HfSe₂ nanosheets to three SF₆ decomposition compounds (SO₂, SO₂F₂, and SOF₂). With biaxial strain, the band gap of the surfaces under discussion expands with tensile up to 1% and gradually decreases for strains between -1% and -5%. Additionally, our findings demonstrate that biaxial strains have a significant impact on the electronic characteristics, adsorption energy (E_{ad}), and desorption time of SO₂, SOF₂, and SO₂F₂ adsorbed HfSe₂ systems. The highest adsorption impact of SO₂, SOF₂, and SO₂F₂ on the bilayer of HfSe₂ occurred at a compressive strain of -5%, with the E_{ad} determined as -1.34, -1.27, and -1.31 eV, respectively. These external factors are widely preferred and offer a workable solution for tunable HfSe₂-based gas sensors and electrical devices under biaxial compressive strain.

Keywords— Strain engineering, Monolayer HfSe₂, Bilayer HfSe₂, SF₆ gas, Gas sensor.

I. INTRODUCTION

To ensure the electricity system's safe operation, SF₆ gas, which has exceptional insulating strength and arc-extinction properties, is frequently used in extra-high voltage equipment [1-2]. However, the inevitably occurring insulating flaws, such as partial discharge and partial overheating during extended use of power equipment, would break down SF₆ over time into several ionized F atoms and low-fluorine sulfides [3]. While these low-fluorine sulfides can often recombine with F atoms to produce SF₆ once more, in the presence of trace amounts of water and oxygen, the sulfides would continue to react with the impure elements to generate a variety of stable gas species, including H₂S, SO₂, SOF₂, and SO₂F₂ [4-5]. These SF₆ decomposition components are dangerous to the environment and human health, speeding up facility corrosion and raising the likelihood of system paralysis [6]. Therefore, operational detection of these typical SF₆ breakdown components is crucial. In this regard, it has been determined that the

detection of SF₆ decomposed species, particularly SO₂, SOF₂, and SO₂F₂, is a practical and efficient method for assessing the operational state of SF₆ insulating devices, which is also significant to ensure the smooth operation of the entire power system [7-8]. Therefore, monitoring the decomposition components, taking precautions promptly, and ensuring the safe operation of power equipment would be possible and practical using gas sensors and adsorbents with ultrahigh-sensitivity and selective qualities [4].

As a result, a wide range of two-dimensional (2D) layered nanosheets, including graphene [9], boron-phosphorus [10], aluminium nitrate [11], aluminum phosphate [12], and silicone [13]. They draw considerable interest due to their distinctive architectures and exceptional characteristics. Due to the structural stability, chemical inertness, and physical properties, the 2D MX₂ family (transition metal dichalcogenides (TMDs); MX₂: M = transition metals (TM); X = chalcogen) has lately demonstrated tremendous potential as gas sensors [14-16]. TMD materials for gas sensing have received a lot of attention, including MoS₂ [17, 18], MoSe₂ [19], PdS₂ [16], PtTe₂ [20], and wS₂ [21]. For example, Li et al. [22] examine and evaluate the outcomes of MoS₂-based gas sensors and concluding that 2D MoS₂ may be suitable for creating high-performance ambient-temperature gas sensors. While other TMD types, such as ZrX₂ and HfX₂ (X = Se, S, and Te), have received less research. HfSe₂ nanosheet has been predicted theoretically to have a small band gap and strong carrier mobility [23]. Numerous scientific experiments have successfully produced the monolayer [24], which suggests that it might be used as a material for gas sensors.

A popular technique for enhancing the nano-system's chemical reaction and catalytic behaviour, and hence enhancing the adsorption and sensing capabilities of the nano-materials, is surface-doping by TM [25-29]. Extensive research on TMDs as SF₆ decomposition sensors, such as MoS₂[30, 31] and MoTe₂ [32]. sensors, has been done during the last few decades. According to the investigations, Ga-MoS₂ might be an excellent gas-sensing material for SO₂ and

SO₂F₂; however, because of its limited adsorption, it was not a good fit for SOF₂ sensing [31]. On the other hand, using first-principles calculations, Cui Hao et al. [33-34] investigated the adsorption behaviour of pristine, and Pt-doped HfSe₂ monolayers to some hazardous gases (SOF₂, SO₂, and NO₂). It is discovered that the Pt-doped HfSe₂ monolayer exhibits better behavior for SO₂. Rh-doped HfSe₂'s adsorption nature and behavior were investigated by Wang et al. [35], and their findings indicate that SO₂ and the monolayer interact more strongly than SO₂F₂. Additionally, our group [36] reported that monolayer HfSe₂ doped with a nickel cluster could distinguish SF₆ decomposition gases, also exhibiting excellent sensitivity and regulated gas release/ storage. Besides, nano-sensors improved sensing capabilities can enable their use in harsher environments with a strong sensitivity toward gaseous species [37-38], which is urgently needed in material science and engineering [39-40]. For adsorption systems, control over charge transfer and electronic structures would impact both sensing performance and adsorption capabilities [41-42].

Applying external strain to the adsorption system is one efficient method for tuning the electronic characteristics [43-44]. For example, Guzman and Strachen [45] forecast significant changes in bandgap for several monolayer TMDs with biaxial strain ranging from -5% to 5%. According to Li et al.'s research [46] The Ru-PtSe₂ monolayer has a tunable sensing response for gas detection with modulated strains, as evidenced by its increased sensitivity to C₂H₂ detection by manipulating biaxial strains. The SnS₂ sheets also demonstrate how biaxial strains can significantly alter the adsorption energy, electronic characteristics, and charge transfer of systems with adsorbed nitrogen dioxide [47]. According to Ni et al.'s findings, the adsorption capabilities of an Ag-doped WSe₂ monolayer declined as the biaxial strain increased [48].

In this study, we design models for SO₂, SOF₂, and SO₂F₂ adsorption on pure monolayer and bilayer HfSe₂ nanosheets to compute the adsorption behaviors, and investigate their potential as gas sensors under applying biaxial strain. Our research may pave the path for using HfSe₂ to identify SF₆ decompositions in challenging application environments, including the interior of gas-insulated switchgear (GIS) equipment.

II. COMPUTATIONAL METHODS

All computations on HfSe₂ nanosheets are performed with DFT via the DMol3 package[49] by using the material studio program, and considering the impact of spin polarization, peripheral electrons for Hf (Se) are produced in a 5d² 6s² (4s²4p⁴) configuration (32 Se atoms and 16 Hf atoms). The generalized approximation gradient (GGA) and Perdew-Burke-Ernzerhof (PBE) are utilized to handle the exchange-related energy [50], where the PBE functional is very popular because it is a non-empirical functional with reasonable accuracy over a wide range of systems. While PBE is typically not the most accurate GGA functional (especially for band gap) for a given system, it usually is not too far off either [51]. In addition, GGA struggles to adequately describe long-range effects like vdWs [52]. To investigate the reactions between HfSe₂ sheets and gas molecules, the long-range dispersion-corrected developed by

Tkatchenko-Scheffler (TS) has been utilized. TS offers empirical parameters for the elements' surface (HfSe₂ sheets), whereas Grimme don't offered it [53]. The basis set was specified by double numerical plus polarization (DNP). We used the appropriate number of Monkhorst-Pack k-point grids, 7 × 7 × 1, and mesh smearing of 0.005 Ha for the geometry optimization and the electronic attribute computations. All structural models were completely relaxed until the maximum force and the tolerance energy were less than 0.002 Ha/Å and 10⁻⁵ eV, respectively. In addition, the self-consistent field (SCF) tolerance was set to 10⁻⁶ Ha [54]. A supercell was large enough for 4 × 4 × 1 HfSe₂ sheets [33, 35, 55]. To avoid erroneous interactions between the periodic pictures, a 20 vacuum was placed between them along the z-axis. The adsorption energy (*E_{ad}*) is defined as the energy required to adsorb a certain amount of gas onto a HfSe₂ sheet [56], which is quantitatively described as follows:

$$E_{ad} = E_{surf/gas} - E_{surf} - E_{gas} \quad (1)$$

where *E_{surf/gas}*, *E_{surf}* and *E_{gas}* are the total energy of gas adsorption systems, isolated surfaces, and the separated gas molecules, respectively. Here, the stronger the adsorption, the more negative the *E_{ads}*. Hirshfeld population analysis can be used to examine the charge transfer (*Q_t*) of the gas molecules adsorbed on layered HfSe₂ sheets [56]. *Q_t* denotes the quantity of charge transfer between substrates and the gas molecules.

$$Q_t = Q_{tiso} - Q_{tads} \quad (2)$$

where *Q_{tiso}* stands for the isolated gas molecules and *Q_{tads}* for the total charges carried of the adsorbed gas.

III. RESULTS AND DISCUSSION

A. The Geometric and Electronic Properties

Fig. 1 (a, b) shows the top and side views of the optimized structures of the HfSe₂ monolayer and bilayer. The P3-m1 space group (No. 164) is home to the tetragonal symmetry found in HfSe₂ sheets. Se, Hf, and Se atoms are stacked one on top of the other to form their structure. The internal coordinates are (0, 0, 0) and (1/3, 2/3, 0.227) for Hf and Se, respectively. The researched material's optimal lattice parameters are *a* = *b* = 3.75 Å, which agree with the theoretical conclusions that have already been established [34, 36]. Interatomic distance Se-Hf is 2.69 Å at equilibrium, and the lattice angles of α= β=90.00° and γ = 120.00°. In our study, AA (eclipsed with Se (Hf) over Se (Hf)) is the order in which the HfSe₂ bilayer was stacked; the interlayer gap between Se and Se is 6.084 Å. The band structure (BS) and density of states (DOS) of the unstrained HfSe₂ monolayer and bilayer are shown in Fig. 1(c, d), respectively. Given that the conduction band (CB) minimum is identified at the M point and the valence band (VB) maximum is discovered at point Γ close to the Fermi level, the profile of the derived BS suggests that the investigated sheets are an indirect semiconductor. The monolayer and bilayer systems' resulting band gaps are 0.477 eV (which agrees with the previously reported value [34-54]) and 0.429 eV, respectively. As observed in the figure, the extent of the energies is limited from -3 eV to 3 eV (-2 eV to 2 eV) for monolayer (bilayer), the VB is dominated mainly by Se atoms, with a modest contribution from Hf atoms, while both atoms contribute to the CB. The response of Se and Hf atoms in the HfSe₂

monolayer and bilayer is reasonable for the hybridization among Se 4p and Hf 5d states, and the hybridization occurs

significantly farther from the Fermi level.

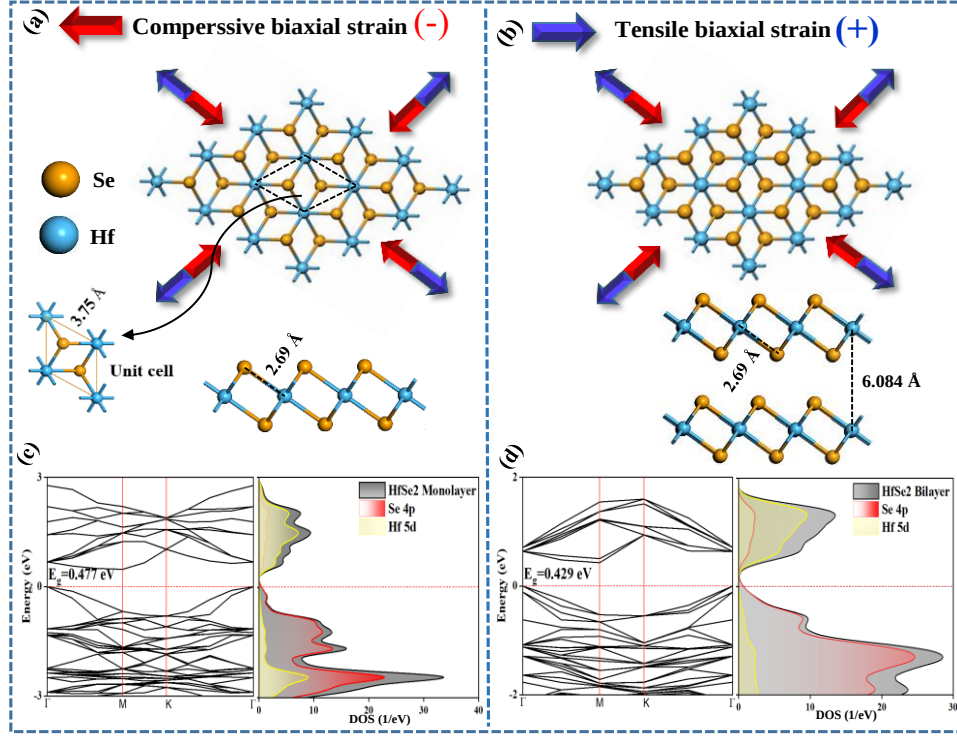


Fig. 1: Geometric structure, BS, and DOS of (a,c) monolayer HfSe₂ and (b,d) bilayer HfSe₂.

B. Adsorption Unstrained Systems

To determine the most favorable adsorption configuration, the three gas molecules were placed molecules at three possible sites above the pristine HfSe₂ monolayer and bilayer positions classified as: 1-the top Se atom of the downer layer, i.e. gas molecules above the center of hexagonal ring of HfSe₂, 2- the top of Hf atom, 3- the top of Se of the upper layer. The three aim gases were adsorbed using the best model of HfSe₂ sheets. The adsorption features of HfSe₂ sheets were investigated in light of the adsorption parameters (represented by E_{ad} , Q_t , and distance between Se and S atoms), which are listed in Table 1. The ideal adsorption models of three gases (SO₂, SOF₂, and SO₂F₂) on the HfSe₂ sheets are shown in Fig. 2. All systems have long adsorption lengths and low Q_t value, indicating insensitive physisorption, as indicated by the values in Table 1.

Table 1. Adsorption characteristics of SO₂, SOF₂, and SO₂F₂ gases on HfSe₂ sheets

Structure	Gas molecules	Figure	E_{ad} (eV)	Q_t Hirshfeld (e)	D (Å) S-Se
HfSe ₂ monolayer	SO ₂	Fig. 2(a1)	-0.32	-0.076	3.194
	SOF ₂	Fig. 2(b1)	-0.247	-0.0008	3.692
	SO ₂ F ₂	Fig. 2(c1)	-0.279	-0.0031	3.866
HfSe ₂ bilayer	SO ₂	Fig. 2(a2)	-0.294	-0.0785	3.173
	SOF ₂	Fig. 2(b2)	-0.214	-0.0015	3.687
	SO ₂ F ₂	Fig. 2(c2)	-0.255	-0.0047	4.03

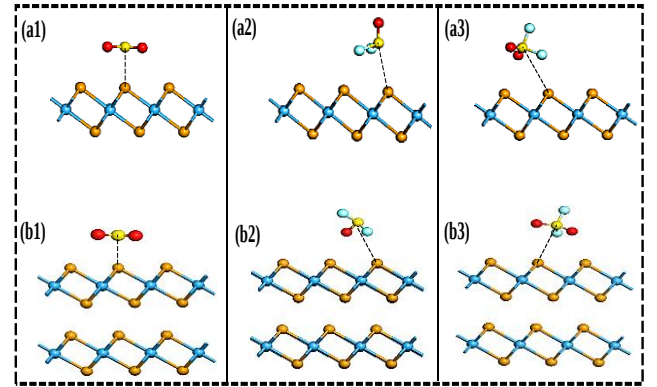


Fig.2: Side view of SO₂, SOF₂ and SO₂F₂ adsorption systems, above (a1-a3) monolayer and (b1-b2) bilayer HfSe₂.

C. Electronic Properties under Strain

Fig. 1 shows the schematic diagrams of applying tensile and compressive biaxial strain on HfSe₂ sheets before gas adsorption, where positive and negative values represent tensile and compressive strain, respectively. Strain engineering is one of the most efficient ways to control the electronic characteristics of semiconductors. It is widely known that altering the electronic characteristics of semiconductor materials is crucial for gas sensors. Biaxial strain is more effective than uniaxial strain in adjusting the adsorption properties [57]. Based on the literature that predicted large band gap changes for several monolayer TMDs with biaxial strain ranging from -5% to 1%. The monolayer shows a transition from semiconductor to metal at -5% with surface stability [45]. Therefore, we examined the

electronic characteristics of the monolayer and bilayer of HfSe₂ while subjected to an external biaxial strain. The lattice constants of the strained and unstrained systems, a and a_0 , respectively, are used to calculate the biaxial strain [56], which is defined as $\varepsilon = (a - a_0)/a_0 \times 100\%$. Only the atomic position can relax for the strained system. Here, strains between -5% and 1% with a step of 2% are taken into consideration; the strain is set with lattice constants of 3.563, 3.637, 3.713, and 3.788 Å, respectively. Three negative/positive strain values correspond to compressive/tensile strains. Analyzing their BSs and DOS prior to gas adsorption is vital to comprehend their adsorption properties under biaxial strain with values -5% , -3% , -1% , and 1% , see Figs. 3 and 4 of monolayer and bilayer, respectively. We discovered that the monolayer and the bilayer have an indirect BG at zero strain and that the gap (decreases) increases at (compressive) tensile uniform strain. The band gaps of the monolayer and bilayer remain indirect for all strain values considered. In the case of a bilayer, the maximum VB changes from Γ to M because the state at M gains energy under tensile strain, although the corresponding eigenvalue at Γ does not significantly change. We can observe that the minimum CB and the maximum VB were always found at the M and Γ point, respectively. Hence, the lack of contrast in maximum VB and the minimum CB governed the non-switch from indirect to direct BG. Keep in mind that negative strain values represent compressive strain. The impact of mechanical distortions on the electronic structure is primarily discernible at the M-K path, where the bottom of the CB is drastically decreased relative to the equilibrium status. The top of VB at the Γ spot travels upward concurrently. Because of the alteration in bond lengths, this behavior is understandable makes sense because the orbital overlap between the Hf and Se atoms reduces. The

electron conductivity in semiconducting TMD nanosheets can be increased or decreased due to changes in the electronic architecture caused by mechanical deformations. The BG is decreased due to the 2D-compressive strain, and the electron conductivity at the Fermi level is noticeably increased. Since the BGs of the HfSe₂ monolayer (bilayer) vanish at about -5% (-3%) strain, the transport paths are open, enabling electrical conductivity. This can be described by the Brillouin zone deformation, where the applied strain evenly alters the primitive cell, bands penetrate the Fermi level, and as a result, a monolayer's (bilayer's) character transforms from semiconducting to metallic. Additionally, it is intriguing to observe that the bilayer's interlayer spacing marginally increases with compressive strain and reduces with tensile strain, as seen in Fig. 5.

D. Enhancement of sensing features under strain

Adsorption systems under biaxial strain ranging from -5% ~ 1% are monitored in order to assess the impact of applied strain on the potential for adsorption and sensing performance of HfSe₂ sheets towards SO₂, SOF₂, and SO₂F₂. The E_{ads} and Q_t with various biaxial strains are depicted in Fig. 6 and 7. The findings demonstrate that applying biaxial strain significantly affects the E_{ad} for all systems, increasing E_{ads} as tensile strain increases. However, as the compressive strain is increased, the adsorption diminishes. In both systems, a better response performance for the sensing material to the target gas is generally correlated with lower E_{ads} . The SO₂F₂ molecule has the same behavior in both systems, but the bilayer exhibits a more considerable sensitivity to the SO₂ and SOF₂ molecules under compressive strains of up to -5% as the E_{ad} gets more negative (-1.34 and -1.27 eV, respectively).

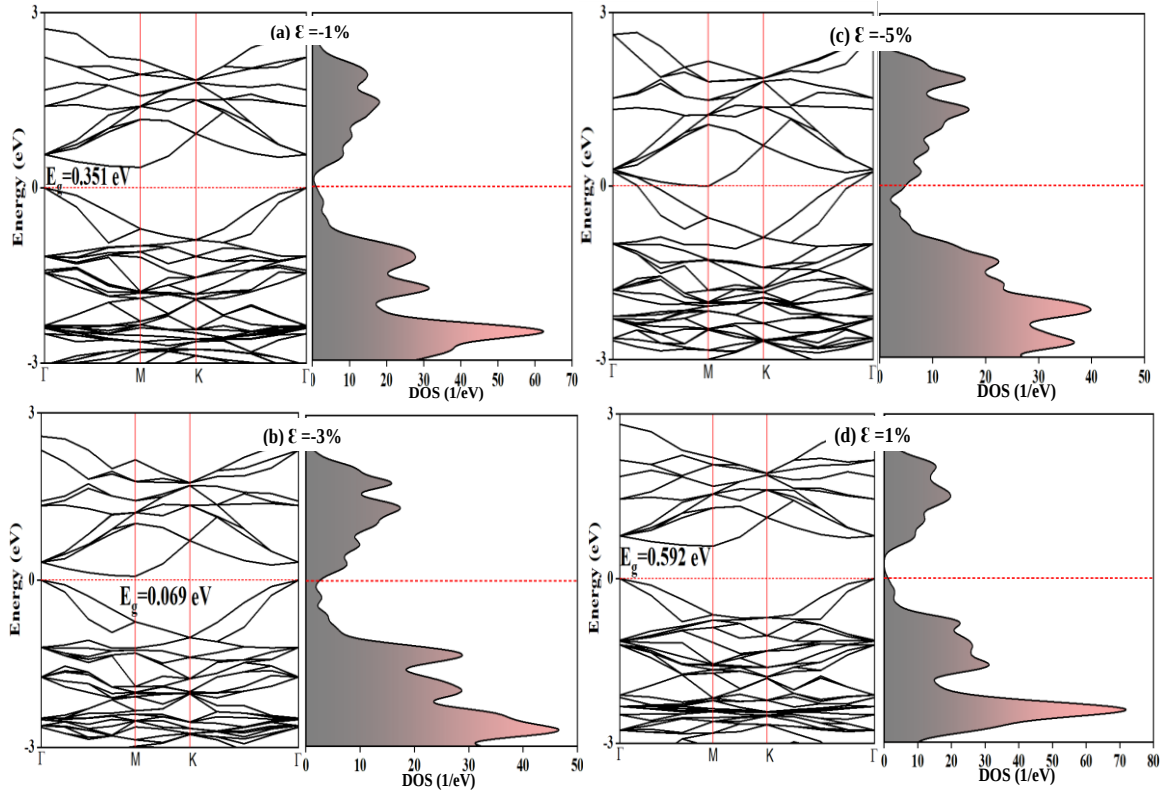


Fig. 3: BGs and DOS for monolayer HfSe₂ in the presence of biaxial strain, represented by the dashed red line at the Fermi level (tuning to zero).

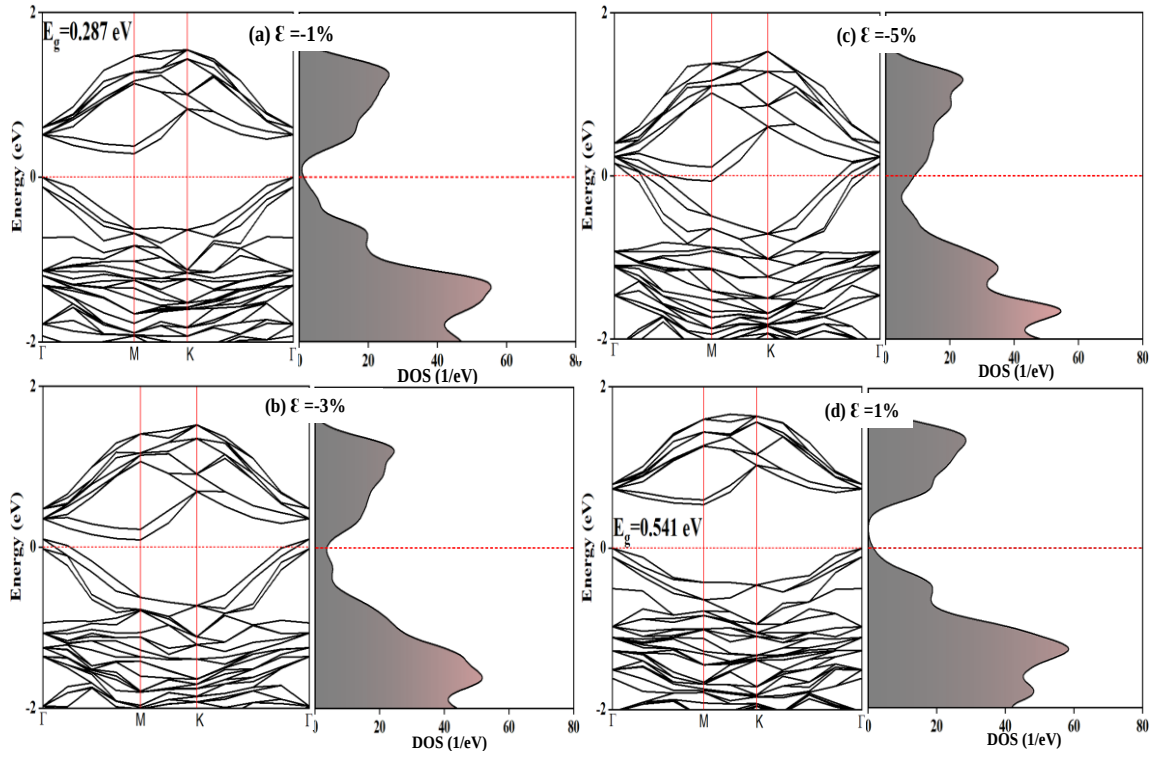


Fig. 4: BGs and DOS for bilayer HfSe₂ in the presence of biaxial strain represented by the dashed red line at the Fermi level (tuning to zero).

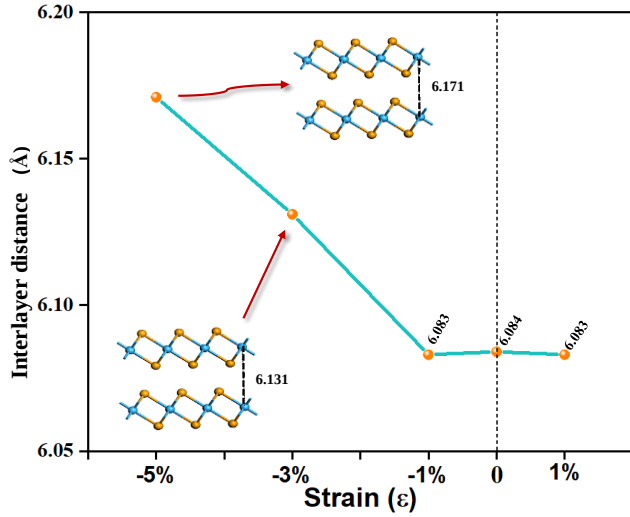


Fig. 5: Interlayer distance for bilayer HfSe₂ as a function of biaxial strain applied, distance in Å.

On the other hand, when the tensile strain reaches 1%, it is expected that the E_{ad} of the HfSe₂ bilayer system will be -0.0803 , $+0.0038$, and -0.0403 eV for SO₂, SOF₂, and SO₂F₂, respectively. This indicates that the adsorption mechanism is weak or non-existent for bilayer systems with a 1% tensile strain. Due to the bilayer system's instability at 1% tensile biaxial strain, the gas molecules are independent of the surface. In other words, the system had some difficulty in adsorption of these gases[31].

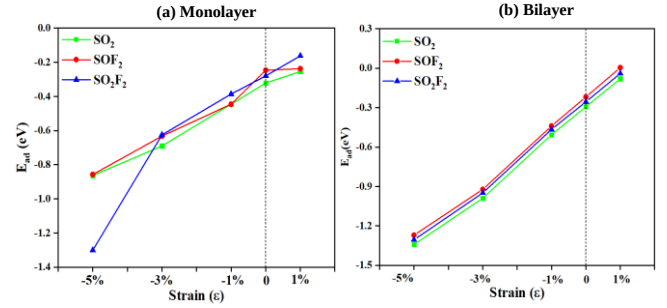


Fig. 6: Variation of the E_{ads} for SO₂, SOF₂, and SO₂F₂ adsorbed on (a) monolayer HfSe₂ and (b) bilayer HfSe₂ as a function of biaxial strain.

In related adsorption systems, the Q_t exhibits a similar tendency as a function of biaxial strain. After adjusting the biaxial strain from -5% to 1% , -0.080 and -0.0787 electrons were transferred by the HfSe₂ monolayer and bilayer from the SO₂ molecule at -5% strain, respectively. Upon examining different systems, it is notable that SO₂-HfSe₂ adsorption systems have the highest Q_t sensitivity. After applying the biaxial strain, there was clear change in the Q_t of the SO₂-HfSe₂ system from -0.08 to $-0.0903e$ for biaxial strain from -5% to 1% , respectively. This result is considered important compared with the charge transfer values in Ref. [58], where Q_t ranges from 0.008 to $0.019e$ for biaxial strain from -4% to 1% , respectively.

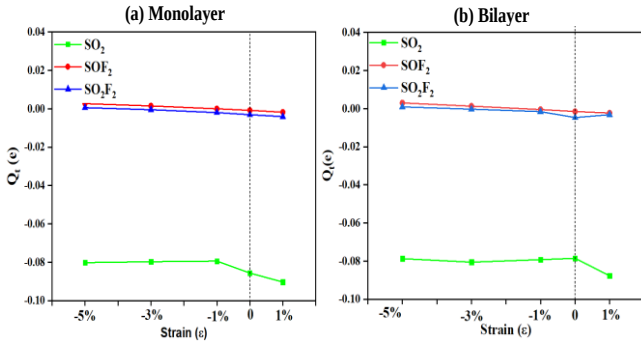


Fig. 7: Variation of the Q_t for SO_2 , SOF_2 , and SO_2F_2 adsorbed on (a) monolayer HfSe_2 and (b) bilayer HfSe_2 as a function of biaxial strain.

In contrast, the SOF_2 and SO_2F_2 adsorption systems (related to low charge transfer) do not experience any substantial Q_t following the addition of external strain. According to Hirshfeld analysis, for SOF_2 -adsorbed monolayer and bilayer HfSe_2 nanosheet at -5% strain, there are 0.0027 e and 0.003 e moving from the substrate to SOF_2 molecule, respectively. The SO_2F_2 adsorption systems behave similarly to SOF_2 systems in Q_t as well. Even if these two gases continue to have the electron donor feature, we might see a change in the charge transport path in the two systems when the applied strain drops below -3% .

In Fig. 8(a and b), we present the relationship between strain and the adsorption distance (d) of HfSe_2 systems. Both systems show slight decrease in the S-Se adsorption distance for the gas molecules with the increase of compression strain (from 0 to -5%), where the absolute value of E_{ad} increases from $0.32 \sim 0.86\text{ eV}$ ($0.29 \sim 1.34\text{ eV}$), $0.247 \sim 0.85$ ($0.22 \sim 1.27\text{ eV}$) and $0.279 \sim 1.3$ ($0.255 \sim 1.31\text{ eV}$) for the adsorption SO_2 , SOF_2 , and SO_2F_2 on HfSe_2 monolayer (bilayer), and a decreasing tendency in the d -curve indicates stronger binding at a reduced adsorption distance. Except for the adsorption of SO_2F_2 gas on the bilayer, we can see that the distance increases and then decreases at -5% strain.

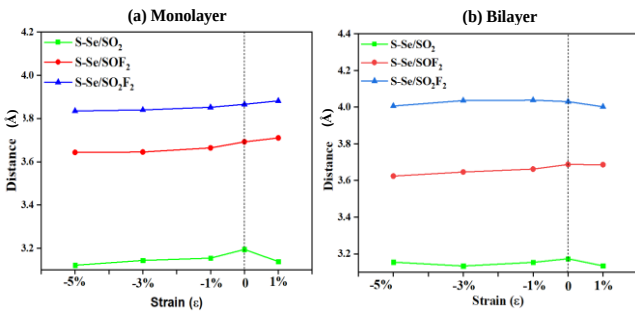


Fig. 8: Variation of the adsorption distance for SO_2 , SOF_2 , and SO_2F_2 adsorbed on (a) HfSe_2 monolayer and (b) HfSe_2 bilayer as a function of biaxial strain.

Initially, we computed the BSs of the studied structures under biaxial strain. The variance in BG with strain is illustrated in Fig. 3 and 4. Generally, compressive strain causes a decrease in the BG of all studied systems. The effect of tensile was slightly different, the BG rising as the biaxial strain increased. The BGs of the gas-adsorbed systems exhibit a linear

relationship in Fig. 9 (a and b), slightly narrower when subjected to compressive strain and broader when tensile strains are applied. In other words, despite both systems being strain-sensitive and giving diverse sensing responses, the HfSe_2 bilayer has greater sensitivity upon gas detection with the modification of biaxial stresses. Adding biaxial strain has less impact the unadsorbed HfSe_2 monolayer, which retains its semiconductor nature except for a compressive strain of up to -5% when the monolayer turns metallic.

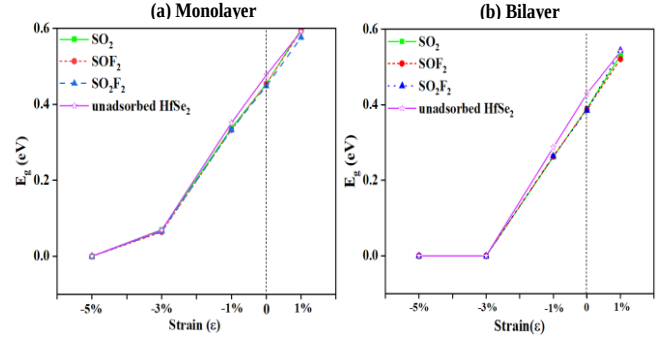


Fig. 9: Variation of the BS for SO_2 , SOF_2 , and SO_2F_2 adsorbed on (a) HfSe_2 monolayer and (b) HfSe_2 bilayer as a function of biaxial strain.

E. Recovery Time

Recovery time (τ), which indicates whether the material is acceptable for practice application reality or not, is another essential merit quality. According to the Arrhenius formula, the relationship between τ and desorption barrier E_a (is equal to E_{ad}) is estimated as follows:

$$\tau = \nu_0^{-1} e^{-E_a/k_B T} \quad (3)$$

Here, ν_0 is the operational frequency, which is taken as 10^{12} Hz , k_B is the Boltzmann constant, and T is the operational temperature (300 K) [59]. The τ is defined by the time it takes the target material (HfSe_2 sheets) to reestablish its pristine properties or condition once the gas molecule is removed, determining how long the HfSe_2 sheets can be reused. Fig. 10(a, b) shows the weighted τ as a function of biaxial strain. According to the calculations shown in Fig. 10, the adsorption time of gas molecules of all systems gets faster (or instantaneously) under the tensile strains and increases linearly under the compressive strains.

For the adsorption on the HfSe_2 monolayer, SO_2F_2 has the largest E_{ad} under the compressive strain of up to -5% , which means that the reaction between SO_2F_2 and the HfSe_2 monolayer is the most powerful. However, the desorption barrier is higher with a high E_{ad} . Since the SO_2F_2 molecule has a much higher estimate of the τ ($9.8 \times 10^9\text{ s}$) than the other two gas molecules, it has the lowest absorption rate. The other two gases have a higher absorption rate due to a lower absorption barrier, and the corresponding τ (6.6 and 5.15 min for SO_2 and SOF_2 gases, respectively) is also shorter.

For the desorption from the HfSe_2 bilayer, all the E_{ads} of these three gas molecules are more significant than those on the HfSe_2 monolayer under the compressive strain. They have higher desorption barriers as a result, which results in slower absorption rates and longer τ . Applying compressive strain significantly lengthens the τ . HfSe_2 sheets have a

more remarkable detection ability when the target gas binds to the surface for a long time. On the other hand, biaxial tensile strain would result in a noticeably reduced τ for each system under consideration, indicating that biaxial tensile strain might be used to desorb gas quickly. For instance, the τ of the gases/HfSe₂ bilayer adsorption system is drastically reduced when the external strain of -5% changes to that of 1%. As a result, the analysis of τ shows that controlling external factors is necessary to regulate the adsorption-absorption process. Due to the combination of the sheets' selectivity of HfSe₂, greater E_{ad} , and the high τ , the bilayer is more effective than the monolayer in detecting SF₆ breakdown products.

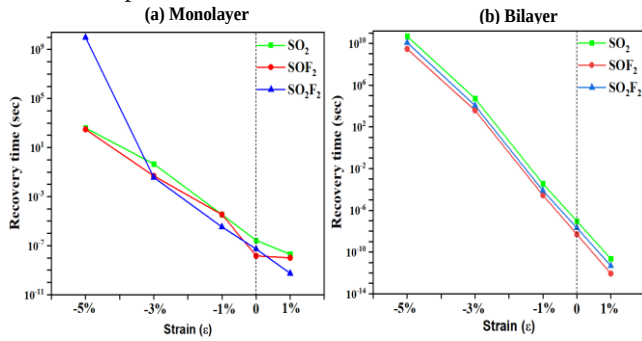


Fig. 10: Variation of the recovery time for SO₂, SOF₂, and SO₂F₂ adsorbed on (a) HfSe₂ monolayer and (b) HfSe₂ bilayer as a function of biaxial strain.

IV. CONCLUSIONS

In brief, the sensor mechanism of layered HfSe₂ towards three common gases resulting from SF₆ decomposition is investigated using first-principles calculations. The geometrical stability and electronic characteristics of SO₂, SOF₂, and SO₂F₂ on the HfSe₂ monolayer and bilayer are monitored under biaxial strain engineering. According to the electronic systems study, biaxial strain could be used to successfully control the electrical conductivities of the surfaces under consideration, while the adsorption of gas molecules does not generate noticeable changes in the electronic properties of the surfaces. However, the E_{ad} of all systems increases after applying compressive strain, while the tensile biaxial strain decreases the E_{ad} . The findings demonstrate that the E_{ads} of SO₂ and SOF₂ molecules on a monolayer (-0.86 and -0.85 eV, respectively) are lower than that for SO₂F₂ adsorption (-1.30 eV) system at -5% strain, indicating that the application of compressive strain increased the HfSe₂ monolayer's sensitivity to SO₂F₂. In addition, the bilayer shows high adsorption energy towards all gases in the following order: SO₂(-1.34) > SO₂F₂(-1.31) > SOF₂(-1.27). Moreover, the computed recovery times for each system showed that tensile strain assisted-absorption, whereas compressive strain enhanced gas adsorption. The study suggested that HfSe₂ nanosheets are a promising bidirectional sensing material for SF₆ decomposition detection, which can be manipulated by applying biaxial strain.

CONFLICT OF INTEREST

Authors declare that they have no conflict of interest.

REFERENCES

- [1] H. Cui, T. Liu, Y. Zhang, and X. Zhang, "Ru-InN monolayer as a gas scavenger to guard the operation status of SF 6 insulation devices: A first-principles theory," *IEEE Sensors Journal*, vol. 19, no. 13, pp. 5249-5255, 2019.
- [2] A. Dong and M. Liu, "A DFT study on the adsorption properties of Ti3C2O2 MXene towards SF6 decomposition gases," *Surface Science*, vol. 734, p. 122317, 2023.
- [3] X. Zhang, L. Yu, Y. Gui, and W. Hu, "First-principles study of SF6 decomposed gas adsorbed on Au-decorated graphene," *Applied Surface Science*, vol. 367, pp. 259-269, 2016.
- [4] Y. Gui, J. Shi, P. Yang, T. Li, C. Tang, and L. Xu, "Platinum modified MoS2 monolayer for adsorption and gas sensing of SF6 decomposition products: A DFT study," *High Voltage*, vol. 5, no. 4, pp. 454-462, 2020.
- [5] F. Zeng, J. Tang, Y. Xie, Q. Zhou, and C. Zhang, "Experiment study of trace water and oxygen impact on SF 6 decomposition characteristics under partial discharge," *Journal of Electrical Engineering & Technology*, vol. 10, no. 4, pp. 1786-1795, 2015.
- [6] A.-J. Yang *et al.*, "Phosphorene: A promising candidate for highly sensitive and selective SF 6 decomposition gas sensors," *IEEE Electron Device Letters*, vol. 38, no. 7, pp. 963-966, 2017.
- [7] X. Zhang, L. Yu, J. Tie, and X. Dong, "Gas sensitivity and sensing mechanism studies on Au-doped TiO2 nanotube arrays for detecting SF6 decomposed components," *Sensors*, vol. 14, no. 10, pp. 19517-19532, 2014.
- [8] H. Zhao *et al.*, "Understanding competitive adsorption of SF6 and its decomposed components on α -Fe2O3," *Surface Science*, vol. 723, p. 122128, 2022.
- [9] Z. Chen, J. Wang, and Y. Wang, "Strategies for the performance enhancement of graphene-based gas sensors: A review," *Talanta*, vol. 235, p. 122745, 2021.
- [10] S. Thomas, V. Kumar, D. R. Roy, and M. A. Zaeem, "Two-dimensional boron-phosphorus monolayer for reversible NO2 gas sensing," *ACS Applied Nano Materials*, vol. 3, no. 10, pp. 10073-10081, 2020.
- [11] W. Abdul-Hussein, F. H. Hanoon, and L. F. Al-Badry, "Control the electronic and optical properties of AlN nanosheet by the electric field," *Results in Optics*, vol. 11, p. 100423, 2023.
- [12] W. Abdul-Hussein, F. H. Hanoon, and L. F. Al-Badry, "Designing sub-5 nm monolayer AlP MOSFETs," *Micro and Nanostructures*, vol. 176, p. 207524, 2023.
- [13] G. K. Walia, D. K. K. Randhawa, and K. S. Malhi, "Rise of silicene and its applications in gas sensing," *Journal of Molecular Modeling*, vol. 27, pp. 1-11, 2021.
- [14] F. Opoku and P. P. Govender, "Highly selective and sensitive detection of formaldehyde by β 12-

- borophene/SnO₂ heterostructures: the role of an external electric field and in-plane biaxial strain," *The Journal of Physical Chemistry A*, vol. 124, no. 11, pp. 2288-2300, 2020.
- [15] C. Niu *et al.*, "Tunable adsorption behavior of small molecule on GeP monolayer by applied strain and electric field," *Applied Surface Science*, vol. 520, p. 146257, 2020.
- [16] Y. Ma, X. Gong, F. Xiao, Y. Liu, and X. Ming, "First-Principles Study on the Selective Detection of Toxic Gases by 2D Monolayer PdS₂: Insight into Charge Transfer Dynamics and Alignment of Frontier Molecular Orbitals," *ACS Applied Nano Materials*, vol. 6, no. 13, pp. 12470-12478, 2023.
- [17] H. Ye *et al.*, "SnSe monolayer: A promising candidate of SO₂ sensor with high adsorption quantity," *Applied Surface Science*, vol. 484, pp. 33-38, 2019.
- [18] L. Liu *et al.*, "High selective gas detection for small molecules based on germanium selenide monolayer," *Applied Surface Science*, vol. 433, pp. 575-581, 2018.
- [19] C. Gao *et al.*, "A DFT study of In doped Ti₂O: a superior NO₂ gas sensor with selective adsorption and distinct optical response," *Applied Surface Science*, vol. 494, pp. 162-169, 2019.
- [20] N. H. Ali and L. F. Al-Badry, "Mo-PtTe₂ monolayer as a promising biosensor for prediagnosis of lung cancer: A DFT study," *Computational and Theoretical Chemistry*, vol. 1230, p. 114379, 2023.
- [21] H. Cui, X. Zhang, G. Zhang, and J. Tang, "Pd-doped MoS₂ monolayer: A promising candidate for DGA in transformer oil based on DFT method," *Applied Surface Science*, vol. 470, pp. 1035-1042, 2019.
- [22] B. Li, Q. Zhou, R. Peng, Y. Liao, and W. Zeng, "Adsorption of SF₆ decomposition gases (H₂S, SO₂, SOF₂ and SO₂F₂) on Sc-doped MoS₂ surface: A DFT study," *Applied Surface Science*, vol. 549, p. 149271, 2021.
- [23] M. Salavati, "Electronic and mechanical responses of two-dimensional HfS₂, HfSe₂, ZrS₂, and ZrSe₂ from first-principles," *Frontiers of Structural and Civil Engineering*, vol. 13, no. 2, pp. 486-494, 2019.
- [24] R. Kumar, N. Goel, M. Hojamberdiev, and M. Kumar, "Transition metal dichalcogenides-based flexible gas sensors," *Sensors and Actuators A: Physical*, vol. 303, p. 111875, 2020.
- [25] Y. Fan, J. Zhang, Y. Qiu, J. Zhu, Y. Zhang, and G. Hu, "A DFT study of transition metal (Fe, Co, Ni, Cu, Ag, Au, Rh, Pd, Pt and Ir)-embedded monolayer MoS₂ for gas adsorption," *Computational Materials Science*, vol. 138, pp. 255-266, 2017.
- [26] D. Ma *et al.*, "The adsorption of CO and NO on the MoS₂ monolayer doped with Au, Pt, Pd, or Ni: A first-principles study," *Applied Surface Science*, vol. 383, pp. 98-105, 2016.
- [27] H. C. Sari *et al.*, "Computational molecular design to assist modification of single-walled carbon nanotubes with B, N, Al, Si, P, and S dopant atoms for Cl₂ gas sensor application," *Surface Science*, vol. 731, p. 122264, 2023.
- [28] R. Zhang, Z. Liu, D. Holiharimanana, and H. Sun, "Renewable Adsorption/Desorption of Sarin on TM-Doped CNTs: First Principle Calculations," *Surface Science*, vol. 736, p. 122338, 2023.
- [29] F. Thamer, A. S. Alwan, and A. B. Ahmed, "Studying the electronic characteristics and physisorption of OTS on the pure silver surfaces (Ag₁₀), (Ag₁₅) and (Ag₁₈)," *University of Thi-Qar Journal of Science*, vol. 10, no. 2, pp. 151-159, 2023.
- [30] E. Piosik and M. J. Szary, "Development of MoS₂ doping strategy for enhanced SO₂ detection at room temperature," 2023.
- [31] W. Hou, H. Mi, R. Peng, S. Peng, W. Zeng, and Q. Zhou, "First-principle insight into Ga-doped MoS₂ for sensing SO₂, SOF₂ and SO₂F₂," *Nanomaterials*, vol. 11, no. 2, p. 314, 2021.
- [32] D.-W. Wang *et al.*, "MoTe₂: A promising candidate for SF₆ decomposition gas sensors with high sensitivity and selectivity," *Ieee Electron Device Letters*, vol. 39, no. 2, pp. 292-295, 2017.
- [33] H. Cui, H. Zhu, and P. Jia, "Adsorption and sensing of SO₂ and SOF₂ molecule by Pt-doped HfSe₂ monolayer: A first-principles study," *Applied Surface Science*, vol. 530, p. 147242, 2020.
- [34] H. Cui, P. Jia, and X. Peng, "Adsorption of SO₂ and NO₂ molecule on intrinsic and Pd-doped HfSe₂ monolayer: A first-principles study," *Applied Surface Science*, vol. 513, p. 145863, 2020.
- [35] X. Wang and Y. Liao, "Selective detection of SO₂ in SF₆ insulation devices by Rh-doped HfSe₂ monolayer: A first-principles study," *Applied Physics A*, vol. 125, pp. 1-8, 2019.
- [36] N. H. Ali and L. F. Al-Badry, "Enhancing Adsorption and Sensing Performances of Pristine and Ni Cluster Doped HfSe₂ Monolayers to SF₆ Decomposition Gases: A DFT Study," *Computational and Theoretical Chemistry*, vol. 1226, p. 114207, 2023.
- [37] D.-H. Baek and J. Kim, "MoS₂ gas sensor functionalized by Pd for the detection of hydrogen," *Sensors and Actuators B: Chemical*, vol. 250, pp. 686-691, 2017.
- [38] F. Xiao, W. Lei, W. Wang, L. Xu, S. Zhang, and X. Ming, "Pentagonal two-dimensional noble-metal dichalcogenide PdS₂ for photocatalytic water splitting with pronounced optical absorption and ultrahigh anisotropic carrier mobility," *Journal of Materials Chemistry C*, vol. 9, no. 24, pp. 7753-7764, 2021.
- [39] W. Lei *et al.*, "A new 2D high-pressure phase of PdSe₂ with high-mobility transport anisotropy for photovoltaic applications," *Journal of Materials Chemistry C*, vol. 7, no. 7, pp. 2096-2105, 2019.
- [40] H. Wu, Y. Xia, C. Zhang, S. Xie, S. Wu, and H. Cui, "Adsorptions of C₅F₁₀O decomposed compounds on the Cu-decorated NiS₂ monolayer:

- a first-principles theory," *Molecular Physics*, vol. 121, no. 3, p. e2163715, 2023.
- [41] Y. Ma, X. Gong, F. Xiao, Y. Liu, and X. Ming, "First-Principles Calculations of Two-Dimensional Monolayer PdSe₂ for Selective Sensing of Nitrogen-Containing Gases," *ACS Applied Nano Materials*, vol. 5, no. 8, pp. 11519-11528, 2022.
- [42] S. Zhai *et al.*, "Single Rh atom decorated pristine and S-defected PdS₂ monolayer for sensing thermal runaway gases in a lithium-ion battery: A first-principles study," *Surfaces and Interfaces*, vol. 37, p. 102735, 2023.
- [43] P. Ou, P. Song, X. Liu, and J. Song, "Superior sensing properties of black phosphorus as gas sensors: a case study on the volatile organic compounds," *Advanced Theory and Simulations*, vol. 2, no. 1, p. 1800103, 2019.
- [44] W. Lei *et al.*, "Ferroelastic lattice rotation and band-gap engineering in quasi 2D layered-structure PdSe₂ under uniaxial stress," *Nanoscale*, vol. 11, no. 25, pp. 12317-12325, 2019.
- [45] D. M. Guzman and A. Strachan, "Role of strain on electronic and mechanical response of semiconducting transition-metal dichalcogenide monolayers: An ab-initio study," *Journal of Applied Physics*, vol. 115, no. 24, 2014.
- [46] D. Li *et al.*, "First-principle insight into the Ru-doped PtSe₂ monolayer for detection of H₂ and C₂H₂ in transformer oil," *ACS omega*, vol. 5, no. 49, pp. 31872-31879, 2020.
- [47] R. Zhao *et al.*, "External electric field and strains facilitated nitrogen dioxide gas sensing properties on 2D monolayer and bilayer SnS₂ nanosheets," *Applied Surface Science*, vol. 491, pp. 128-137, 2019.
- [48] J. Ni, W. Wang, M. Quintana, F. Jia, and S. Song, "Adsorption of small gas molecules on strained monolayer WSe₂ doped with Pd, Ag, Au, and Pt: A computational investigation," *Applied Surface Science*, vol. 514, p. 145911, 2020.
- [49] B. Delley, "From molecules to solids with the DMol 3 approach," *The Journal of chemical physics*, vol. 113, no. 18, pp. 7756-7764, 2000.
- [50] B. Zhu *et al.*, "Monolayer Janus Te₂Se-based gas sensor to detect SO₂ and NO_x: a first-principles study," *Physical Chemistry Chemical Physics*, vol. 23, no. 2, pp. 1675-1683, 2021.
- [51] J. P. Perdew, K. Burke, and M. Ernzerhof, "Generalized gradient approximation made simple," *Physical review letters*, vol. 77, no. 18, p. 3865, 1996.
- [52] P. Ziesche, S. Kurth, and J. P. Perdew, "Density functionals from LDA to GGA," *Computational materials science*, vol. 11, no. 2, pp. 122-127, 1998.
- [53] A. Tkatchenko and M. Scheffler, "Accurate molecular van der Waals interactions from ground-state electron density and free-atom reference data," *Physical review letters*, vol. 102, no. 7, p. 073005, 2009.
- [54] N. H. Ali and L. F. Al-Badry, "Improvement of sensing performance of SO₂, SOF₂, and SO₂F₂ gases on monolayer and bilayer HfSe₂ under electric field," *Micro and Nanostructures*, vol. 182, p. 207654, 2023.
- [55] H. Wu, B. Zhang, X. Li, and X. Hu, "First-principles screening upon Pd-doped HfSe₂ monolayer as an outstanding gas sensor for DGA in transformers," *Computational and Theoretical Chemistry*, vol. 1208, p. 113553, 2022.
- [56] H. Yang *et al.*, "Strain-engineered S-HfSe₂ monolayer as a promising gas sensor for detecting NH₃: A first-principles study," *Surfaces and Interfaces*, vol. 34, p. 102317, 2022.
- [57] J. Cao *et al.*, "Controllable gas sensitive performance of 1T'WS₂ monolayer instructed by strain: First-principles simulations," *Chemical Physics Letters*, vol. 758, p. 137921, 2020.
- [58] Z. Xu, "Ni-Decorated PtS₂ Monolayer as a Strain-Modulated and Outstanding Sensor upon Dissolved Gases in Transformer Oil: A First-Principles Study," *ACS omega*, vol. 8, no. 6, pp. 6090-6098, 2023.
- [59] V. Kumar, A. Bano, K. Rajput, and D. R. Roy, "The interaction of two-dimensional P₂SiS nanosheet with environmental toxic NCG molecules for sensor application: A DFT study," *Sensors and Actuators A: Physical*, vol. 322, p. 112608, 2021.